Enhancement of the electric dipole moment of the electron in diatomic molecules YbF and

## HgF

M. G. Kozlov. Zh. Exp. Teor. Fiz. 89, 1933 (1985)
M. G. Kozlov, V. F. Ezhov. Phys. Rev. A, 49, 4502 (1994)

## SEMI-EMPIRICAL WAVE FUNCTION

The main assumption of the method is that the WF of the molecule can be written as follows

$$
\begin{aligned}
& |\Lambda, \Omega\rangle= \\
& |\lambda=\Lambda, \omega=\Omega\rangle_{\text {unpaired }}|\Lambda=0, \Omega=0\rangle_{\text {coupled }}
\end{aligned}
$$

In the vicinity of the heavy nucleus ( Hg or Yb ) the WF can be expanded in spherical waves

$$
|\lambda, \omega\rangle=\sum_{l, j(=l \pm 1 / 2)} C_{l, j}|l, j, \omega\rangle
$$

Here $|l, j, \omega\rangle$ are normalized four-component spherical waves:

$$
|l, j, \omega\rangle=\binom{f_{l, j} Y_{j, \omega}^{l}}{i g_{l, j} Y_{j, \omega}^{l^{\prime}}}
$$

$f$ and $g$ are radial functions, $Y_{j, \omega}^{l}$ is the spherical spinor, $l^{\prime}=2 j-l$.

At the small distances solutions of the Dirac equation are:

$$
\binom{f_{l, j}}{g_{l, j}}=\frac{k}{|k|} \frac{a_{l}}{Z^{1 / 2} r}\binom{(\gamma+k) J_{2 \gamma}(x)-\frac{x}{2} J_{2 \gamma-1}(x)}{\alpha Z J_{2 \gamma}(x)}
$$

where

$$
\begin{aligned}
x & =\sqrt{8 Z r} \\
\gamma & =\sqrt{(j+1 / 2)^{2}-\alpha^{2} Z^{2}}, \\
k & =(l-j)(2 j+1)
\end{aligned}
$$

For each $l \neq 0$ a pair of functions with $j=l-1 / 2$ and $j=l+1 / 2$ on the large distances have to form nonrelativistic function $\left|l, m_{l}=\lambda, \omega\right\rangle$. It means, that

$$
\begin{aligned}
& C_{1,3 / 2}=-\sqrt{2} C_{1,1 / 2} \\
& C_{2,5 / 2}=-\sqrt{3 / 2} C_{2,3 / 2}, \text { etc. }
\end{aligned}
$$

For small $r$ molecular WF is defined by the products

$$
\sigma_{s}=C_{0,1 / 2} a_{0}, \quad \sigma_{p}=C_{1,1 / 2} a_{1}, \quad \text { etc. }
$$

The hyperfine axial tensor $\mathbf{A}$ :

$$
\begin{aligned}
A= & \frac{A_{\|}+2 A_{\perp}}{3} \\
= & \frac{4}{3} C_{-1}^{2} h_{-1,-1}+\frac{4}{9} C_{1}^{2} h_{1,1}+\frac{8}{9} C_{-2}^{2} h_{-2,-2} \\
& +\frac{8}{15} C_{2}^{2} h_{2,2}+\frac{4}{5} C_{-3}^{2} h_{-3,-3} \\
& -\frac{8 \sqrt{2}}{9} C_{1} C_{-2} h_{1,-2}-\frac{8 \sqrt{6}}{15} C_{2} C_{-3} h_{2,-3} \\
A_{d}= & \frac{A_{\|}-A_{\perp}}{3} \\
= & -\frac{8}{9} C_{1}^{2} h_{1,1}-\frac{8}{45} C_{-2}^{2} h_{-2,-2} \\
& -\frac{8}{15} C_{2}^{2} h_{2,2}-\frac{8}{35} C_{-3}^{2} h_{-3,-3} \\
& -\frac{2 \sqrt{2}}{9} C_{1} C_{-2} h_{1,-2}-\frac{2 \sqrt{6}}{15} C_{2} C_{-3} h_{2,-3}
\end{aligned}
$$

Single index $k$ is used instead of $l, j$;

$$
h_{i, j}=-\frac{g_{n} \alpha}{2 m_{p}} \int_{0}^{\infty}\left(f_{i} g_{j}+g_{i} f_{j}\right) d r
$$

$g_{n}$ is $G$-factor of the nucleus; $m_{p}$ is the proton mass.

Using experimental values for $A$ and $A_{d}$ we obtain equations for parameters $\sigma_{i}$. For YbF molecule these equations have the form:

$$
\left\{\begin{aligned}
31170 \sigma_{s}^{2}-1330 \sigma_{p}^{2}-50 \sigma_{d}^{2}+\cdots & =7617 \\
5510 \sigma_{p}^{2}+370 \sigma_{d}^{2}+\cdots & =102
\end{aligned}\right.
$$

The right hand side correspond to the experimental values for $A$ and $A_{d}$ in MHz measured by Van Zee et al.

Coefficients $C_{l, j}$ are normalized to unity. The semi-classical analysis gives $a_{l}^{2} \approx 0.3$. It results in approximate normalization condition

$$
\sigma_{s}^{2}+3 \sigma_{p}^{2}+\frac{5}{2} \sigma_{d}^{2} \leq 0.3 \sum_{l, j} C_{l, j}^{2}=0.3
$$

The solution which meets this condition correspond to the negligible contribution of the $d$-wave (less then $5 \%$ ).

$$
\begin{aligned}
& \sigma_{s}^{2}=0.24, \quad \sigma_{p}^{2}=0.019 \\
& \sigma_{s} \sigma_{p}=-0.067
\end{aligned}
$$

## SPIN-ORBIT INTERACTION

Mixing of the ground state with the first excited state $(\lambda=1, \omega=1 / 2):$

$$
\xi=\frac{\langle\lambda=1, \omega| H_{s o}|\lambda=0, \omega\rangle}{E_{0}-E_{1}}
$$

For YbF that is equal to

$$
\xi=0.37 \frac{\langle 6 p, \lambda=1, \omega| H_{s o}|6 p, \lambda=0, \omega\rangle}{18000 \mathrm{~cm}^{-1}} \approx-0.03
$$

For HgF molecule this mixing is only slightly larger. It means, that the accuracy of the equations

$$
\begin{aligned}
& C_{1,3 / 2}=-\sqrt{2} C_{1,1 / 2} \\
& C_{2,5 / 2}=-\sqrt{3 / 2} C_{2,3 / 2}, \text { etc. }
\end{aligned}
$$

is about $3 \%$.

## $P$ - AND $T$-ODD INTERACTIONS

Interaction of the EDM of the electron $d_{e}$ with the molecular electric field $(-\nabla \phi)$ :

$$
H_{d}=2 d_{e}\left(\begin{array}{cc}
0 & 0 \\
0 & \vec{\sigma}
\end{array}\right)(-\nabla \phi)
$$

$$
\langle\lambda, \omega| H_{d}|\lambda, \omega\rangle=W_{d} d_{e} \omega
$$

$$
\begin{aligned}
& W_{d}=8 \sum_{i} C_{-i} C_{i} \int_{0}^{\infty} g_{-i} g_{i} \frac{d \phi}{d r} r^{2} d r \\
& =16 \alpha^{2} Z^{3}\left(\frac{\sigma_{s} \sigma_{p}}{\gamma_{1 / 2}\left(4 \gamma_{1 / 2}^{2}-1\right)}-\frac{\sqrt{2} \sigma_{p} \sigma_{d}}{\gamma_{3 / 2}\left(4 \gamma_{3 / 2}^{2}-1\right)}\right)
\end{aligned}
$$

For YbF molecule this expression gives

$$
W_{d}=174 \sigma_{s} \sigma_{p}-15 \sigma_{p} \sigma_{d}+\ldots=-11.7
$$

Electron-nuclear scalar interaction

$$
H_{S}=i \frac{G \alpha}{\sqrt{2}} Z k_{S} \gamma_{0} \gamma_{5} n(\mathbf{r})
$$

$G$ is the Fermi constant; $\gamma_{i}$ are the Dirac matrices; $n(\mathbf{r})$ is the nuclear density normalized to unity. Dimensionless factor $Z k_{S}$ is: $Z k_{S}=Z k_{S, p}+N k_{S, n}$.

Interaction of the nuclear magnetic quadrupole moment $M$ with the molecular magnetic field:

$$
\begin{aligned}
H_{M} & =-\frac{M}{4 I(2 I-1)} T_{i, k} \frac{3}{2 r^{5}} \gamma_{0} \gamma_{j} r_{l}\left(\epsilon_{j, l, i} r_{k}+\epsilon_{j, l, k} r_{i}\right) \\
T_{i, k} & =I_{i} I_{k}+I_{k} I_{i}-\frac{2}{3} I(I+1) \delta_{i, k}
\end{aligned}
$$

$P$-odd, $T$-even interaction of the nuclear anapole moment with valence electron:

$$
H_{A}=\frac{G \alpha}{\sqrt{2}} k_{A} \mathbf{I} \gamma_{0} \vec{\gamma} n(\mathbf{r})
$$

$k_{A}$ is the anapole moment constant for the ytterbium nucleus (Khriplovich et al).

## SPIN-ROTATIONAL HAMILTONIAN

$$
\begin{aligned}
H_{s r}= & B \mathbf{N}^{2}+\gamma \mathbf{S} \cdot \mathbf{N}+\mathbf{S} \cdot \mathbf{A I}+\mathbf{S} \cdot \mathbf{A}^{\prime} \mathbf{I}^{\prime} \\
& +\mu_{0} \mathbf{S} \cdot \mathbf{G B}-D \mathbf{n} \cdot \mathbf{E} \\
& +W_{A} k_{A} \mathbf{n} \times \mathbf{S} \cdot \mathbf{I}+\left(W_{S} k_{S}+W_{d} d_{e}\right) \mathbf{S} \cdot \mathbf{n} \\
& +\frac{-1}{4 I(2 I-1)} T_{i, k}\left(\frac{1}{2} Q q_{0} n_{i} n_{k}+2 M W_{M} S_{i} n_{k}\right)
\end{aligned}
$$

In this expression $\mathbf{I}$ and $\mathbf{I}^{\prime}$ are the spins of Yb and F nuclei; $\mathbf{N}$ is the rotational angular momentum; $B$ and $\gamma$ are the rotational and the spin-doubling constants. Tensors $\mathbf{A}$ and $\mathbf{A}^{\prime}$ correspond to the hyperfine structures on two nuclei. $\mu_{0}$ is the Bohr magneton; $\mathbf{n}$ is the molecular axis unit vector directed from Yb (or Hg ) to F .

Experimental values for $H_{s r}$ parameters for ${ }^{171} \mathrm{Yb}^{19} \mathrm{~F}$ molecule:

$$
\begin{array}{ll}
B=7237 M H z ; \\
A_{\|}=7822 M H z, & A_{\perp}=7513 M H z(I=1 / 2) ; \\
A_{\|}^{\prime}=220 M H z, & A_{\perp}^{\prime}=134 M H z\left(I^{\prime}=1 / 2\right) ; \\
G_{\|}=1.9975, & G_{\perp}=1.9954
\end{array}
$$

Constant $\gamma$ is found from the relation (Knight et al)

$$
\gamma=-2 B\left(G_{\perp}-G_{\|}\right) \approx 30 M H z
$$

The rough estimate for $D$ :

$$
\begin{aligned}
D & =-R_{0}+2 C_{0,1 / 2} C_{1,1 / 2}\langle 6 s| r|6 p\rangle \\
& \approx-5.2 \cdot 10^{-18} C G S E
\end{aligned}
$$

$R_{0}$ is the internuclear distance.
Constants $q_{0}$ and $W_{M}$ do not vanish for ${ }^{173} \mathrm{Yb}(I=5 / 2)$. If $Q=2.8 \cdot 10^{-24} \mathrm{~cm}^{2}$, then

$$
Q q_{0}^{v a l}=1880 M H z
$$

Values of the constants $W_{i}$ are as follows:

$$
\begin{aligned}
W_{A} & =0.73 \mathrm{kHz} \\
W_{S} & =-48 \mathrm{kHz} \\
W_{d} & =-1.5 \cdot 10^{25} \frac{\mathrm{~Hz}}{e c m} \\
W_{M} & =2.1 \cdot 10^{33} \frac{\mathrm{~Hz}}{e e^{2}}
\end{aligned}
$$

## BEST EXPERIMENTAL LIMITS ON THE

## CONSTANTS OF $P$ - AND $T$-ODD INTERACTIONS

 AND CORRESPONDING FREQUENCY SHIFTS IN MOLECULES| Const. | Group | Year | Object | Upper bound | Frequency shift (Hz) |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  | TlF | HgF | YbF |
| $Q_{S}$ | Seatle | 1993 | ${ }^{199} \mathrm{Hg}$ | $3 \cdot 10^{-50} \mathrm{ecm}^{3}$ | $5 \cdot 10^{-5}$ | - | - |
| $C_{T}$ | " | " | " | $2 \cdot 10^{-8}$ | $3 \cdot 10^{-5}$ | - | - |
| $C_{S}$ | " | " | " | $1 \cdot 10^{-6}$ | $1 \cdot 10^{-4}$ | 0.2 | 0.05 |
| M | Amherst | 1989 | ${ }^{133} \mathrm{Cs}$ | $1 \cdot 10^{-33} \mathrm{ecm}^{2}$ | - | 1.5 | 0.7 |
| $d_{e}$ | Berkeley | 1990 | Tl | $1 \cdot 10^{-26}$ e cm | - | 0.5 | 0.15 |

